

Green Energy 2021: Photo assisted methanation using Cu_2O nanoparticles supported on graphene as photocatalyst

Josep Albero

Universitat Politècnica de València (UPV), Spain

Abstract

Photoassisted CO_2 methanation can be carried out efficiently at 250 °C using Cu_2O nanoparticles supported on few layer graphene ($\text{Cu}_2\text{O}/\text{G}$) as photocatalyst. The $\text{Cu}_2\text{O}/\text{G}$ photocatalyst has been prepared by chemical reduction of a Cu salt ($\text{Cu}(\text{NO}_3)_2$) with ethylene glycol in the presence of defective graphene obtained from the pyrolysis of alginate acid at 900 °C under Ar flow. Using this photocatalyst a maximum specific CH_4 formation rate of 14.93 mmol/g $\text{Cu}_2\text{O}\cdot\text{h}$ and apparent quantum yield of 7.84 % was achieved, which is one of the highest reported values for the gas-phase methanation reaction at temperatures below Sabatier reaction (>350 °C). It was found that the most probable reaction mechanism involves photoinduced electron transfer from the $\text{Cu}_2\text{O}/\text{G}$ photocatalyst to CO_2 , while evidence indicates that light-induced local temperature increase and H_2 activation are negligible. The role of the temperature in the process has been studied, the available data suggesting that heating is needed to desorb the H_2O formed as product during the methanation. The most probable reaction mechanism seems to follow dissociative pathway involving detachment of oxygen atoms from CO_2 .